

Supporting document 1

Risk and technical assessment report

Application A1088 – Sodium Hydrosulphite as a Food Additive

# Executive Summary

FSANZ received an Application from Seafood New Zealand Limited seeking to amend Standard 1.3.1 – Food Additives of the *Australia New Zealand Food Standards Code* (the Code) to permit the addition of sodium hydrosulphite as a bleaching agent in canned abalone.

Sulphur dioxide and six sulphite compounds are currently permitted as food additives in canned abalone at a maximum permitted level (MPL) of 1000 mg/kg, calculated as sulphur dioxide. For the proposed use of sodium hydrosulphite in canned abalone, this Application requests an identical MPL of 1000 mg/kg, calculated as sulphur dioxide.

The food technology assessment concluded that sodium hydrosulphite fulfils the stated technological function as a bleaching agent at the proposed level of use. The Application indicated that sodium hydrosulphite is the most effective compound available to produce a canned abalone product with acceptable organoleptic properties.

During the processing of canned abalone, sodium hydrosulphite undergoes chemical decomposition to produce the same chemical species that result from use of the other approved sulphites, namely hydrogen sulphite (HSO3–), sulphite (SO32–), hydrogen sulphate (HSO4–), sulphate (SO42–) and sulphur dioxide (SO2). No residual sodium hydrosulphite is detectable in the final canned product. Therefore, the use of sodium hydrosulphite in the production of canned abalone will not result in dietary exposure to a new food additive nor additional dietary exposure to sulphites.

FSANZ is currently conducting a risk assessment of sulphites in the Australia New Zealand food supply as part of Proposal P298 - Benzoate and sulphite permissions in food. The sulphite permissions for canned abalone are not under review in this Proposal because consumption of canned abalone is very low compared to foods that are the major contributors to total dietary exposure to sulphites.

Some individuals are sensitive to sulphites (e.g. asthmatics) and this will also be the case for sodium hydrosulphite. The Code requires the mandatory declaration of sulphites in the statement of ingredients when added to foods in concentrations of 10 mg/kg or more.

It is concluded that the use of sodium hydrosulphite as a food additive in canned abalone is technologically justified and presents no identifiable public health and safety issues above those of the sulphite food additives currently permitted in canned abalone.

#

# Table of Contents

[Executive Summary 1](#_Toc386529518)

[Table of Contents 2](#_Toc386529519)

[1. INTRODUCTION 3](#_Toc386529520)

[1.1 Risk assessment context 3](#_Toc386529521)

[1.2 Risk and Technical Assessment questions 3](#_Toc386529522)

[2. FOOD TECHNOLOGY ASSESSMENT 3](#_Toc386529523)

[2.1 Characterisation of sodium hydrosulphite 4](#_Toc386529524)

[2.1.1 Identity 4](#_Toc386529525)

[2.1.2 Chemical and physical properties 4](#_Toc386529526)

[2.1.3 Production of sodium hydrosulphite 5](#_Toc386529527)

[2.1.4 Specifications 5](#_Toc386529528)

[2.1.5 Methods of analysis in foods 5](#_Toc386529529)

[2.2 Technological function of sodium hydrosulphite 5](#_Toc386529530)

[2.3 Conclusion 5](#_Toc386529531)

[3. HAZARD ASSESSMENT 6](#_Toc386529532)

[4. DIETARY EXPOSURE ASSESSMENT 6](#_Toc386529533)

[5. RISK CHARACTERISATION 6](#_Toc386529534)

[6. RISK AND TECHNICAL ASSESSMENT CONCLUSIONS 6](#_Toc386529535)

[6.1 Responses to risk and technical assessment questions 6](#_Toc386529536)

[6.2 Consolidated conclusion 7](#_Toc386529537)

[7. REFERENCES 7](#_Toc386529538)

# Introduction

On the 7 June 2013, Food Standards Australia New Zealand (FSANZ) received an Application from Seafood New Zealand Limited seeking to amend Standard 1.3.1 – Food Additives of the *Australia New Zealand Food Standards Code* (the Code) to permit the addition of sodium hydrosulphite to canned abalone as a bleaching agent.

Standard 1.3.1 currently permits the following food additives for use in canned abalone at a maximum permitted level (MPL) of 1000 mg/kg, calculated as sulphur dioxide: sulphur dioxide (INS no. 220), sodium sulphite (221), sodium bisulphite (222), sodium metabisulphite (223), potassium metabisulphite (224), potassium sulphite (225) and potassium bisulphite (228). For the proposed use of sodium hydrosulphite in canned abalone, this Application requests an identical MPL of 1000 mg/kg, calculated as sulphur dioxide.

## 1.1 Risk assessment context

For the purpose of this risk assessment, the proposed addition of sodium hydrosulphite to canned abalone in Australia and New Zealand will be considered in the context of the following:

* Sodium hydrosulphite is proposed as an alternative to sulphites currently approved for use in canned abalone;
* The proposed MPL of 1000 mg/kg, expressed as sulphur dioxide, is identical to the existing MPL for sulphites.

1.2 Risk and Technical Assessment questions

For this Application, the risk assessment questions were developed in the context of the Section 18 Objectives of the *Food Standards Australia New Zealand Act 1991*.

The following risk assessment questions are addressed in this report:

1. Does sodium hydrosulphite achieve its technological function in the form and quantity used as a bleaching agent for canned abalone?

2. Are there any public health and safety issues associated with the use of sodium hydrosulphite as a bleaching agent for canned abalone?

# 2. Food technology assessment

Sodium hydrosulphite is proposed as an alternative food additive to be used in the processing of canned New Zealand abalone to bleach the natural black colour to the more commercially acceptable golden blonde to nutmeg colour, especially for the export market. Sodium hydrosulphite has preservative, antioxidant and bleaching properties. It is not currently permitted for this purpose in the Code, nor is it a Codex Alimentarius permitted food additive. However, it is permitted to perform a similar function in other country’s food regulations. The Code permits a number of other sulphite food additives for treating canned abalone, but it is claimed they do not bleach the colour of New Zealand abalone to a commercially acceptable extent compared to sodium hydrosulphite.

## 2.1 Characterisation of sodium hydrosulphite

### 2.1.1 Identity

Common name: Sodium hydrosulphite

Other names: Sodium dithionite; disodium hydrosulphite; dithionous acid, disodium salt; sodium hyposulphite; sodium sulfoxilate

C.A.S. registry number: 7775-14-6

INS number: Not Applicable (does not have an official Codex Alimentarius food additive INS number)

Structural formula: Na2S2O4

Molecular weight: 174.1 g/mol

Molecular structure:



### 2.1.2 Chemical and physical properties

Sodium hydrosulphite is a white solid powder which is very soluble in water, where it decomposes. It is a strong reducing agent.

Dobson (2011) stated that canned abalone has a pH in the range 6.0 – 6.5, and in this pH range the majority of free sulphite is in the form of the sulphite anion (SO32-), with a minor proportion as the bisulphite anion (HSO3-). Sulphur dioxide (SO2) does not occur in solution in this pH range, and is only found in sulphite solutions when the pH is below 4.0 and becomes the predominant form when the pH is < 2.0 (Green 1976).

An unpublished report provided in the Application indicated that the hydrosulphite anion (S2O42-) which results from the initial dissolution of sodium hydrosulphite in aqueous solution was not detectable in the final canned abalone product (Harding et al 2011).

The chemical and physical properties of sodium hydrosulphite are provided in the following table.

|  |  |
| --- | --- |
| **Property** | **Value** |
| Physical state at room temperature  | White powder |
| Melting point | Decomposes above 90°C |
| Boiling point | Not applicable |
| Relative density | 2.38 g/cm3 (at 20°C) |
| Vapour pressure | Not applicable |
| Water solubility | Approx. 182 g/L (at 20°C) |
| Partition coefficient n-octanol/water (log value) | Less than -4.7 |

###

### 2.1.3 Production of sodium hydrosulphite

Since sodium hydrosulphite is relatively unstable it is usually stabilised with sodium carbonate and sodium sulphite, which are generally permitted food additives, at about 20%. Both of these extra food additives are already permitted to be added to canned abalone, so this would not raise any extra regulatory issue.

### 2.1.4 Specifications

There is no specification for sodium hydrosulphite in either of the primary sources of specifications in Standard 1.3.4 – Identity and Purity (i.e. not in the JECFA (Joint FAO/WHO Expert Committee on Food Additives) Combined Compendium of Food Additive Specifications, nor the Food Chemicals Codex). However, there is a specification for sodium hydrosulfite (alternative spelling) in the Japanese Specifications and Standards for Food Additives (7th Edition, 2000) which is a secondary source of specifications in clause 3 of Standard 1.3.4. Therefore, no specification for sodium sulphite is required to be written into the Schedule for Standard 1.3.4. Sodium hydrosulphite is permitted as a food additive in Japan to treat a range of different foods.

### 2.1.5 Methods of analysis in foods

Official methods based on the original Monier-Williams method are appropriate for the determination of sulphites in canned abalone (e.g. Hillery et al 1989).

## 2.2 Technological function of sodium hydrosulphite

Sodium hydrosulphite has the technological function as a preservative, antioxidant and bleaching agent like the other sulphiting agents which are already permitted in the Code for addition to various food categories including canned abalone.

Sodium hydrosulphite is effective for bleaching the black colour of native New Zealand abalone (*Haliotis iris* or “paua”) to a more commercially desirable golden blonde to nutmeg colour. The hydrosulphite reacts during the treatment to produce sulphur dioxide as the final by-product which will have the function as a preservative and antioxidant in the canned product, like the other currently permitted sulphites.

Treatment of abalone with sodium hydrosulphite occurs prior to canning. The Application explains that other sulphiting agents (INS 220, 221, 222, 223, 224, 225 and 228) which are already permitted in the Code to treat canned abalone are not as effective as bleaching agents. Massey University in New Zealand was commissioned to investigate alternative reducing and bleaching agents to sodium hydrosulphite but the researcher concluded that there were no suitable alternatives in terms of product quality, colour, flavour and texture in the resulting canned abalone product. FSANZ has no reason to disagree with this conclusion.

## 2.3 Conclusion

Sodium hydrosulphite is concluded to be appropriate as a food additive with function as a preservative, antioxidant and bleaching agent to treat native New Zealand abalone (paua) to bleach the black colour to a more consumer acceptable golden blonde to nutmeg colour, especially for the international market.

# 3. Hazard assessment

The use of sodium hydrosulphite as a food additive for the production of canned abalone only results in the formation of chemical species that are also produced from the use of sulphites that are currently approved for use in canned abalone.

FSANZ has conducted a hazard assessment of sulphites as part of Proposal P298 - Benzoate and sulphite permissions in food (FSANZ 2005a). The original papers used by the Joint FAO/WHO Expert Committee on Food Additives (WHO 1974) to derive a group Acceptable Daily Intake (ADI) for sulphur dioxide and sulphites, expressed as sulphur dioxide, together with a number of additional papers which have been published since the ADI was established, were reviewed by FSANZ. It was concluded that there was no new suitable evidence available that could change the existing group ADI of 0–0.7 mg/kg bw, expressed as sulphur dioxide.

Sulphite consumption has been linked with allergy-like symptoms and asthmatic reactions in individuals who are sensitive to sulphites (Simon 1998; FSANZ 2005b) and this will also be the case for sodium hydrosulphite.

# 4. Dietary exposure assessment

A dietary exposure assessment was not conducted because the proposed use of sodium hydrosulphite will not result in additional dietary exposure to sulphites.

Regarding Proposal P298, the sulphite permissions for canned abalone are not under review because consumption of canned abalone is very low compared to the foods that are the largest contributors to dietary exposure to sulphites (sausages, dried apricots and cordial, in the case of children, and white wine, sausages and dried apricots for adults: FSANZ 2005b).

# 5. Risk characterisation

Sodium hydrosulphite undergoes chemical decomposition to produce the same chemical species that result from the use of existing approved sulphites. The use of sodium hydrosulphite in the production of canned abalone will not result in dietary exposure to a new food additive nor additional dietary exposure to sulphites. The use of sodium hydrosulphite as a food additive in canned abalone presents no identifiable public health and safety issues above those of the sulphite food additives currently permitted in canned abalone.

# 6. Risk and technical assessment conclusions

This risk and technical assessment evaluated the technological suitability and safety of the proposed addition of sodium hydrosulphite to canned abalone.

6.1 Responses to risk and technical assessment questions

### 1. Does sodium hydrosulphite achieve its technological function in the form and quantity used as a bleaching agent for canned abalone?

|  |  |
| --- | --- |
| ***Section of report*** | ***Summary response/conclusion*** |
| Section 2 | Evidence submitted in support of this Application provides adequate assurance that sodium hydrosulphite fulfils the stated technological function as a bleaching agent for canned abalone. |

### 2. Are there any public health and safety issues associated with the use of sodium hydrosulphite as a bleaching agent in canned abalone?

|  |  |
| --- | --- |
| ***Section of report*** | ***Summary response/conclusion*** |
| Sections 3 and 5 | Sulphites are associated with adverse reactions in some individuals, however there are no additional identifiable public health and safety issues associated with the proposed use of sodium hydrosulphite as a bleaching agent for canned abalone. |

6.2 Consolidated conclusion

It is concluded that the use of sodium hydrosulphite as a food additive in canned abalone is technologically justified and presents no additional identifiable public health and safety issues above that of the sulphite food additives currently permitted in canned abalone.

# 7. REFERENCES

Dobson S (2011) Risk assessment of sulphites in Australian canned abalone. Australian Seafood Cooperative Research Centre. Project No. 2008/729. November 2011. Available at:

<http://safefish.com.au>

FSANZ (2005a) Proposal P298 - Benzoate and sulphite permissions in food. Initial Assessment Report. Food Standards Australia New Zealand. August 2005. Available at:

<http://www.foodstandards.gov.au/code/proposals/Pages/proposalp298benzoate2973.aspx>

FSANZ (2005b) 21st Australian Total Diet Study - A total diet study of sulphites, benzoates and sorbates. Food Standards Australia New Zealand. August 2005. Available at:

<http://www.foodstandards.gov.au/publications/Pages/21staustraliantotald2963.aspx>

Green LF (1976) Sulphur dioxide and food preservation – a review. *Food Chemistry*, **1**:103-124.

Harding DRK, Kavianinia I, Ji SML (2011) Tracking the fate of sodium dithionite during the bleaching of New Zealand black abalone – paua. Institute of Fundamental Sciences, Massey University, Palmerston North, New Zealand. Unpublished report.

Hillery BR, Elkins ER, Warner CR, Daniels D, Fazio T, Balazs P, Bosquez MH, Chaddha R, Cordes S, Couture K, et al (1989) Optimized Monier-Williams method for determination of sulfites in foods: collaborative study. *J Assoc Off Anal Chem*, **72**(3):470-475.

OECD (2004) Screening Information Data Set (SIDS) Initial Assessment Report on sodium dithionite. Organisation for Economic Co-operation and Development. Available at:

<http://www.chem.unep.ch/irptc/sids/OECDSIDS/7775146.pdf>

Simon RA (1998) Update on sulfite sensitivity. Allergy, **53**(Suppl 46):78-79.

WHO (1974) Sulfur dioxide and sulfites. Joint FAO/WHO Expert Committee on Food Additives meeting, Geneva, 25 June – 4 July, 1973, WHO Food Additives Series No. 5.